

New cyclohexylidenehydrazide and 4-aza-1-thiaspiro[4.5]decan-3-one derivatives of 3-phenyl-4(3H)-quinazolinones

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Abstract

In this study, the synthesis of (3-phenyl-4(3H)-quinazolinon-2-yl)mercaptoacetic acid cyclohexylidenehydrazides and 4-[(3-phenyl-4(3H)-quinazolinon-2-yl)mercaptoacetyl amino]-4-aza-1-thiaspiro[4.5]decan-3-ones and the results of the study on their antifungal activity are reported. Most of the tested compounds were found to be active against *Microsporum gypseum* NCPF-580, *Microsporum canis*, *Tricophyton mentagrophytes* NCPF-375 var. *erinacei* and *Tricophyton rubrum* at 25 µg/ml. © 1998 Elsevier Science S.A. All rights reserved.

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1. Introduction

Since the discovery of 4(3H)-quinazolinones and their important clinical properties, there has been intense effort to find new compounds of this type which are more potent, more selective and, therefore, without the concomitant effects associated with these drugs. Among the different structural variations carried out on these molecules, the main and more fruitful trends were those based on the change in the 4(3H)-quinazolinone nucleus. Some of the compounds synthesized following these criteria (methaqualone, phenquizone, etc.) have exceeded the clinical trials and are used at present as hypnotic-sedative and diuretic agents. Furthermore, it has been shown that 4(3H)-quinazolinonylhydrazides and related heterocyclic systems could be considered as antimicrobials [1,2]. On the other hand, the presence of the thiazole ring in compounds with a wide range of biological activities has also contributed to the enlargement of interest in the closely related 4-thiazolidones which are reported to display useful antimicrobial properties [3,4]. These observation prompted us to synthesize cyclohexylidenehydrazides containing the 4(3H)-quinazolinone nucleus and their cyclization products, spirothiazolidinones, which may possess antifungal activity.

2. Experimental

2.1. Chemistry

Melting points were determined with a Büchi 530 apparatus in open capillaries and are uncorrected. IR spectra were recorded on KBr discs, using a Perkin-Elmer 1600 spectrometer. ¹H NMR spectra were run on a Bruker AC 200 (200 MHz) or Bruker DPX 400 (400 MHz) spectrometer. ¹³C NMR spectra were recorded on a Bruker AC 200 (50.3 MHz) spectrophotometer. EI/MS were determined on a VG Zab Spec (70 eV) or a VG-Platform (70 eV). Elemental analyses were performed on a Carlo Erba 1106 elemental analyzer. Analyses indicated by the symbols were within ±0.4% of the theoretical values.

2.1.1. Synthesis of (3-phenyl-4(3H)-quinazolinon-2-yl)-mercaptoacetic acid cyclohexylidenehydrazides 2a–c

Cyclohexanone (0.01 mol) was added to a solution of (3-phenyl-4(3H)-quinazolinon-2-yl)mercaptoacetic acid hydrazides **1** (0.01 mol) in absolute ethanol. The reaction mixture was refluxed over a water bath for 4 hours and allowed to stand overnight. The white needles formed were filtered and recrystallized from absolute ethanol.

The following abbreviations are used: cyclohex. = cyclohexane, quin. = quinazolinone, thiaz. = thiazolidone, arom. = aromatic.

Spectral data of **2a**. IR (ν cm⁻¹, KBr): 3236 (NH), 1687, 1667 (C=O). ¹H NMR (200 MHz, δ ppm, DMSO-d₆): 1.58

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(s, 6H, cyclohex.), 2.21, 2.36 (2s, 4H, cyclohex. 2-CH₂, 6-CH₂), 3.98, 4.35 (2s, 2H, SCH₂), 7.44–8.01 (m, 8H, arom.), 10.41, 10.48 (2s, 1H, NH). ¹³C NMR (APT) (50.3 MHz, δ ppm, DMSO-d₆): 24.91 (s, cyclohex. C₄), 25.49, 26.73 (2s, cyclohex. C₃, C₅), 27.22, 34.75 (2s, cyclohex. C₂, C₆), 35.05, 35.22 (2s, SCH₂), 120.72 (s, quin. C_{4a}), 125.38 (s, quin. C₈), 127.95 (s, phenyl C₄), 129.16 (s, phenyl C₂, C₆), 129.41 (s, phenyl C₃, C₅), 129.93 (s, quin. C₅), 134.77 (s, quin. C₇), 135.43 (s, phenyl C₁, quin. C₆), 145.73 (s, quin. C_{8a}), 156.60 (s, quin. C₂), 157.95 (s, cyclohex. C=N), 159.57 (quin. CO), 162.95, 168.59 (2s, amide CO). EI/MS [m/z (% relative intensity)]: 440 [M^+ , 6 (442, 2)], 329 [100 (331, 47)], 287 [82 (289, 44)].

Spectral data of **2c**. IR (ν cm⁻¹, KBr): 3215 (NH), 1690, 1667 (C=O). ¹H NMR (400 MHz, δ ppm, DMSO-d₆): 1.00, 1.02 (2d, J =6.4, 6.5 Hz, 3H, CH₃), 1.02–2.42 (m, 8H, cyclohex.), 2.95 (q, 1H, cyclohex. 4-CH), 3.40, 4.08 (2s, 2H, SCH₂), 7.56–7.71 (m, 6H, quin. 8-CH, C₆H₅), 7.94, 7.97 (2dd, J =2.4, 8.7 Hz; J =2.4, 8.7 Hz, 1H, quin. 7-CH), 8.11 (t, J =3.1 Hz, 1H, quin. 5-CH), 10.56, 10.63 (2s, 1H, NH). EI/MS [m/z (% relative intensity)]: 454 [M^+ , 0.3 (456, 0.1)], 329 [2 (331, 0.7)], 287 [3 (289, 1)], 77 (100).

2.1.2. Synthesis of 4-[(3-phenyl-4(3H)-quinazolinon-2-yl)-mercaptoacetyl amino]-4-aza-1-thiaspiro[4.5]decan-3-ones **3a–c** and **4a,b**

A mixture of **2** (0.005 mol), thioglycolic acid or thiolactic acid (0.01 mol) in dry benzene (30 ml) was refluxed for 48 hours by a water separator which removes the liberated water. Benzene was removed by distillation under reduced pressure. The oil thus obtained was washed with 10% sodium bicarbonate solution in order to remove the excess acid. The precipitate was filtered, washed with water and recrystallized from absolute ethanol.

Spectral data of **3a**. IR (ν cm⁻¹, KBr): 3224 (NH), 1738, 1694 (C=O). ¹H NMR (200 MHz, δ ppm, DMSO-d₆): 1.22 (t, J =7.1 Hz, 1.5H, 0.5CH₃CH₂OH), 1.38–1.74 (m, 10H, cyclohex.), 3.54 (s, 2H, SCH₂), 3.99 (s, 2H, thiaz. SCH₂), 4.14 (q, J =7.1 Hz, 1H, 0.5CH₃CH₂OH), 7.45–8.03 (m, 9H, arom.), 10.29 (s, 1H, NH).

Spectral data of **3b**. IR (ν cm⁻¹, KBr): 3237 (NH), 1727, 1684 (C=O). ¹H NMR (200 MHz, δ ppm, DMSO-d₆): 1.45–1.70 (m, 10H, cyclohex.), 3.53, 3.99 (2s, 2H, SCH₂), 3.66 (s, 2H, thiaz. SCH₂), 7.36–8.02 (m, 8H, arom.), 10.36 (br s, 1H, NH). ¹³C NMR (APT) (50.3 MHz, δ ppm, DMSO-d₆): 22.69 (s, cyclohex. C₄), 23.84 (s, cyclohex. C₃, C₅), 27.64 (s, cyclohex. C₂, C₆), 33.98, 34.25 (2s, SCH₂), 36.63 (s, thiaz. SCH₂), 72.01 (s, spiro C), 120.76 (s, quin. C_{4a}), 125.23 (s, quin. C₈), 127.86 (s, phenyl C₄), 128.87 (s, quin. C₅), 129.11 (s, phenyl C₂, C₆), 129.37 (s, phenyl C₃, C₅), 134.63 (s, quin. C₇), 135.67 (s, phenyl C₁, quin. C₆), 145.53 (s, quin. C_{8a}), 156.75 (s, quin. C₂), 158.13 (s, quin. CO), 167.29 (s, amide CO), 182.17 (s, thiaz. CO). EI/MS [m/z (% relative intensity)]: 514 [M^+ , 10 (516, 4)], 329 [100 (331, 40)], 287 [67 (289, 37)].

Spectral data of **3c**. IR (ν cm⁻¹, KBr): 3228 (NH), 1725, 1677 (C=O). ¹H NMR (200 MHz, δ ppm, DMSO-d₆): 0.86 (d, J =5.6 Hz, 3H, CH₃), 1.09–2.50 (m, 9H, cyclohex.), 3.56 (s, 2H, thiaz. SCH₂), 3.64, 3.73 (2s, 2H, SCH₂), 7.24–8.01 (m, 8H, arom.), 10.18 (br s, 1H, NH). EI/MS [m/z (% relative intensity)]: 528 [M^+ , 10 (530, 4)], 329 [97 (331, 46)], 287 [100 (289, 52)].

Spectral data of **4a**. IR (ν cm⁻¹, KBr): 3241 (NH), 1732, 1685 (C=O). ¹H NMR (200 MHz, δ ppm, DMSO-d₆): 1.19–1.72 (m, 13H, CH₃, cyclohex.), 3.68, 4.11 (2s, 2H, SCH₂), 3.84 (q, J =5.1 Hz, 1H, thiaz. SCH), 7.16–8.24 (m, 9H, arom.), 10.18, 10.33 (2s, 1H, NH).

Spectral data of **4b**. IR (ν cm⁻¹, KBr): 3235 (NH), 1729, 1682 (C=O). ¹H NMR (200 MHz, δ ppm, DMSO-d₆): 1.24–1.72 (m, 13H, CH₃, cyclohex.), 3.69, 4.12 (2s, 2H, SCH₂), 3.86 (q, J =4.8 Hz, 1H, thiaz. SCH), 7.24–8.02 (m, 8H, arom.), 10.18, 10.38 (2s, 1H, NH). ¹³C NMR (APT) (50.3 MHz, δ ppm, DMSO-d₆): 19.53 (s, CH₃), 22.58 (s, cyclohex. C₄), 23.01, 23.85 (2s, cyclohex. C₃, C₅), 25.18, 34.31 (2s, cyclohex. C₂, C₆), 36.52 (s, SCH), 36.76, 37.70 (2s, SCH₂), 70.69 (s, spiro C), 120.59 (s, quin. C_{4a}), 125.37 (s, quin. C₈), 128.07 (s, phenyl C₄), 128.78 (s, quin. C₅), 129.15 (s, phenyl C₂, C₆), 129.45 (s, phenyl C₃, C₅), 134.71 (s, quin. C₇), 135.54 (s, phenyl C₁, quin. C₆), 145.89 (s, quin. C_{8a}), 156.83 (s, quin. C₂), 159.87 (s, quin. CO), 169.72 (s, amide CO), 180.34 (s, thiaz. CO). EI/MS [m/z (% relative intensity)]: 528 [M^+ , 10 (530, 4)], 329 [100 (331, 43)], 287 [50 (289, 25)].

2.2. Microbiology

2.2.1. Antifungal activity

All the compounds to be tested were dissolved in DMSO at a concentration of 4000 μ g/ml and the final concentration was reduced to 200 μ g/ml with sterile distilled water. No effect of DMSO (5%) was observed upon growth of dermatophytes. The dermatophyte strains which were grown on slant medium of Sabouraud (Difco) were transferred to 3.5 ml nutrient broth (NB, Diagnostic Pasteur) and incubated for three to five days at 25°C. At the end of the incubation period these strains were transferred into screwcapped bottles containing sterilized beads and shaken for 4–5 min in a vortex (IKA-VF, Germany). The suspensions of the cultures were adjusted to have an absorbance degree of 0.6 at 450 nm in the spectrophotometer. Eight different dilutions between 25 and 0.2 μ g/ml were prepared in microplates by serial dilutions from top to bottom. Then all the wells except the 12th wells (positive control) were filled with 10 μ l of the standardized strains. These plates were incubated at 25°C for 5 or 6 days.

The minimum concentration at which no growth was observed was taken as the MIC value. It should be noted, however, that these techniques leave a variable number of broken hyphae and therefore even an identical optical density of such hyphal suspensions could lead to a considerable var-

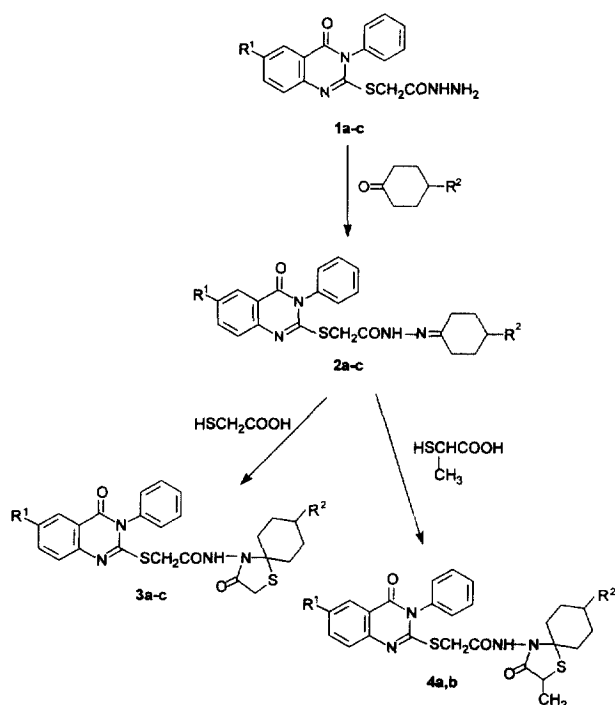
iation in the number of viable cells; this would obviously prevent proper standardization of the inoculum.

3. Results and discussion

The reaction of (3-phenyl-4(3*H*)-quinazolinon-2-yl)-mercaptoacetic acid hydrazides **1** [5] with cyclohexanone afforded the corresponding (3-phenyl-4(3*H*)-quinazolinon-2-yl)mercaptoacetic acid cyclohexylidenehydrazides **2a–c** [6]. 4-[(3-Phenyl-4(3*H*)-quinazolinon-2-yl)mercaptoacet-

ylamino]-4-aza-1-thiaspiro[4.5]decan-3-ones **3a–c** and 4-[(3-phenyl-4(3*H*)-quinazolinon-2-yl)mercaptoacetylmino]-4-aza-2-methyl-1-thiaspiro[4.5]decan-3-ones **4a,b** were synthesized by treatment of **2** with thioglycolic acid and thiolactic acid in dry benzene, respectively (Scheme 1). The structures of the synthesized compounds were confirmed by physical (Table 1) and spectral (IR, ¹H NMR, ¹³C NMR, EI/MS) data.

The IR spectra of **2**, **3** and **4** showed two separate bands resulting from the NH and CO bands of the amide function at about regions 3241–3194 and 1694–1655 cm^{−1}, respectively [1]. The lactam CO band was observed at about 1694–1677 cm^{−1} [7]. A new CO band at 1738–1725 cm^{−1} in the spectra of **3** and **4** provided evidence for the spirothiazolidone structure [8,9]. In the ¹H NMR spectra of **2**, **3** and **4** the SCH₂ (δ 3.54 or 3.53–3.98 and 3.73–4.35 ppm) and NH protons (δ 10.18–10.36 or 10.18–10.41 and 10.33–10.48 ppm) of the mercaptoacetylmino moiety were observed as a singlet or a double singlet presumably due to the partial double bond character of the C–N bond and the bulk of the attached cyclohexyl or 1-thia-4-azaspiro[4.5]decan-3-one structure which can disrupt free rotation about the cited bond [10]. The spectra displayed the SCH₂ (δ 3.56–3.99 ppm) and SCH (δ 3.84–3.86 ppm) resonances as a singlet and quartet, confirming cyclization to **3** and **4**, respectively [9,11]. In the APT ¹³C NMR spectra of **2a**, **3b** and **4b** chosen as prototypes, all the carbons resonated in the expected regions [11–14]. The spectra of **3b** and **4b** did not display the cyclohexylidene C=N peak, whereas they displayed the SCH₂ (δ 36.63 ppm), spiro C (δ 72.01 ppm) and C=O (δ 182.17 ppm) peaks in **3b** and the CH₃ (δ 19.53 ppm), SCH (δ 36.52 ppm), spiro C (δ 70.69 ppm) and C=O (δ 180.34 ppm) peaks in **4b**, which verified the proposed spirothiazolidone structures [11,14]. The EI/MS spectra of **2a**, **2c**, **3b**, **3c** and **4b** showed molecular ions (*M*⁺) with low intensity and two common fragmentation routes, which was

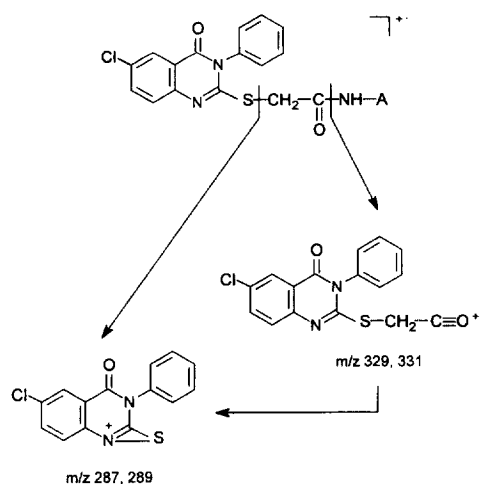


Scheme 1. Synthesis of **2a–c**, **3a–c** and **4a,b**.

Table 1
Physical constants of compounds **2**, **3** and **4**

Comp.	R ¹	R ²	Yield (%)	M.p. (°C)	Formula (mol. wt.)
2a	Cl	H	92	226–230	C ₂₂ H ₂₁ ClN ₄ O ₂ S (440.95)
2b	CH ₃	CH ₃	84	> 300	C ₂₄ H ₂₆ N ₄ O ₂ S · 1.5H ₂ O (461.58)
2c	Cl	CH ₃	79	173	C ₂₃ H ₂₃ ClN ₄ O ₂ S (454.97)
3a ^a	H	H	82	125	C ₂₄ H ₂₄ N ₄ O ₃ S ₂ · 0.5C ₂ H ₅ OH (503.63)
3b	Cl	H	75	195	C ₂₄ H ₂₃ ClN ₄ O ₃ S ₂ · 2H ₂ O (551.07)
3c	Cl	CH ₃	86	146	C ₂₅ H ₂₅ ClN ₄ O ₃ S ₂ · 3H ₂ O (583.13)
4a	H	H	81	106	C ₂₅ H ₂₆ N ₄ O ₃ S ₂ · 2H ₂ O (530.67)
4b	Cl	H	77	130	C ₂₅ H ₂₅ ClN ₄ O ₃ S ₂ · 2.5H ₂ O (574.11)

^a The ethanol peak was observed in the ¹H NMR spectrum.



Scheme 2. Mass fragmentation routes of compounds 2, 3 and 4.

Table 2
MIC values ($\mu\text{g/ml}$) of compounds 2, 3 and 4

Comp.	Fungi ^a			
	A	B	C	D
2a	> 25	25	25	25
2b	25	25	> 25	25
2c	> 25	25	25	25
3a	> 25	25	> 25	> 25
3b	25	25	25	25
3c	25	25	–	25
4a	> 25	> 25	> 25	> 25
4b	25	25	25	25
Ketoconazole	0.2	0.4	0.8	1.6

^a A = *Microsporum gypseum* NCPF-580, B = *Microsporum canis*, C = *Trichophyton mentagrophytes* NCPF-375 var. *erinacei*, D = *Trichophyton rubrum*.

consistent with the literature [5]. In the first route the $CO-NH$ linkage was broken, yielding the peak at m/z 329 (331) which was the base peak in most cases. In the second route, fragment ion at m/z 287 (289) was formed by the cleavage of the $S-CH_2$ bond (Scheme 2).

Compounds 2, 3 and 4 were evaluated for in vitro antifungal activity against *Microsporum gypseum* NCPF 580, *Microsporum canis*, *Trichophyton mentagrophytes* var. *erinacei* NCPF 375 and *Trichophyton rubrum* using the microdilution method [15]. Ketoconazole was used as the standard in the tests. As can be seen in Table 2, most of the tested compounds showed inhibition against the tested fungi at 25 $\mu\text{g/ml}$. The preliminary results indicated that R^1 - and/or R^2 -substituted compounds were more active than entries which were unsubstituted. 3b ($R^1 = \text{Cl}$, $R^2 = \text{H}$) was the most potent compound and was effective against all fungi, whereas 4a ($R^1 = \text{H}$, $R^2 = \text{H}$) was devoid of antifungal activity.

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